Evaluation of water quality of River Malin using Water Quality Index (WQI) at Najibabad, Bijnor (UP) India

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31 Author contribution

SN	Name of the author	Contribution	Sign		
1	Rakesh Bhutiani	Supervision, Formal analysis, review editing, and Plagiarism	- Low		
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3	Faheem Ahamad	Conceptualization, investigation, original drafts preparation	-fateen Adamad		

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35 Abstract (Maximum 300 words)

Malin river originates from the Kotdwara and merges with Ganga at RavalliGhat in Bijnor. It 36 is an important river of city Najibabad Distt- Bijnor (U.P) because it is the main source of 37 irrigation for agriculture in the most areas of city Najibabad. At some places cloth washing 38 and vegetable washing is the main activity on the Malin River bank. Four sampling sites were 39 established for the collection of water samples during July, 2015 to June, 2016 but in the 40 present study average of all the values of all the four sites was given. Monitoring of water of 41 River Malin includes physico-chemical parameters like temperature, turbidity, total solids, 42 total suspended solids, total dissolved solids, pH, total hardness, calcium hardness, 43 magnesium hardness, total alkalinity, chloride, acidity, dissolved oxygen, biochemical 44 45 oxygen demand and chemical oxygen demand. TDS, total hardness, calcium hardness and magnesium hardness was found beyond the limit at all the four sampling sites and rest all the 46 parameters were found within the limit. The average values of TDS, BOD, COD and TH 47 were observed 635.1 mg/l±55.31, 12.1±0.54, 35.2±1.01, 341.0±1.84. Further water quality of 48 river Malin has been assessed using water quality index and the quality of river Malin was 49 observed to be bad at all site which may be attributed to untreated and/or partially treated 50 waste inputs of municipal and industrial effluents joining the river. 51

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Key words: BOD, COD, Malin River, Non-perennial, Ravalli Ghat, WQI (in alphabetic
 order)

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56 Introduction

Rivers form the lifeline of human society and play an important role in the development of 57 Nation and sustenance of life, which are being polluted due to rapid industrialization, 58 urbanization and other developmental activities (Mandal et al., 2012; Aalam and Pathak 59 2010; Mandal and Das 2011). These are vital freshwater systems of strategic importance 60 across the world, providing main water resources for domestic, industrial, agricultural and 61 recreational purposes. Most of the agriculture area in India receives its water from surface 62 sources like river, reservoir, dam etc. River may be perennial as well as non-perennial. In 63 perennial rivers water flows for all the seasons because such rivers are snow fed. The non-64 perennial rivers get dried in summer either partially or completely and in monsoon, they are 65 flooded with water. Generally the quantity of water available from non-perennial rivers varies 66 throughout the year. It normally decreases in summer when demand for water is at its 67 maximum. The Malin River under study is also a non-perennial river. 68

Insufficient capacity of waste water treatment and increasing sewage generation pose big question of disposal of waste water. This huge quantity of waste water is directly and after partial treatment discharged into nearby water bodies mainly in the rivers. The river under study was also heavily polluted due to sewage and industrial discharge (Bhutiani and Ahamad, 2018).

Controlling water pollution is urgent for ecological sustainability of water resources as well 74 75 as for underlying economic reasons and human health. The availability of good quality water is an indispensable feature for preventing diseases and improving quality of life. It is 76 necessary to know information about different physico-chemical parameters before it is used 77 78 for different purposes (Kolhe and Shinde, 2014). The term water quality was developed to 79 give an indication of how suitable the water is for human consumption (Vaux, 2001) and is widely used in multiple scientific publications related to the necessities of sustainable water 80 management (Parparov et al., 2006). 81

Therefore a regular monitoring of river water quality not only prevents outbreak of diseases and checks water from further deterioration, but also provides a scope to assess the current investments for pollution prevention and control (Sudevi and Lokesh, 2012).

85

86 The WQI was first developed by Horton in the early 1970s. The basic aim of WQI is to give a single value to the water quality of a source on the basis of one or the other system which 87 translates the list of constituents and their concentrations present in a sample in to a single 88 value (Abbasi and Abbasi, 2012). The index result represents the level of water quality in a 89 given water basin, such as lake, river or stream. After Horton a number of workers all over 90 the world developed WQI based on rating of different water quality parameters. For the 91 92 evaluation of water quality, WQI was applied to the river water (Singh, 1992; Naik and Purohit, 2001; Kumar and Dua, 2009; Kumar et al., 2009; Sharma et al., 2009; Singkran et 93 94 al., 2010; Gupta et al., 2012). In the present paper, characteristics of different point sources contributing Malin river are discussed and water quality of river Malin is assessed using 95 96 water quality index.

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99 Material and Methods

100 Analysis of water was performed according to standard methods as prescribed by APHA (1998), Trivedy and Goel (1986) and Khanna and Bhutiani (2011) for the examination of the 101 water and waste water. In the present study the samples were collected three times in a month 102 103 in morning hours (7 am-10 am) from July 2015 to June 2016 from Malin River Najibabad Distt Bijnor (UP). Grab water samples from sites were collected in plastic jerry cans from 104 105 about 15 cm below the surface water by keeping and opening Jerri cans below the surface water. Caps of cans were removed after dipping the can and also closed in the water after 106 filling up of can. Care was taken to avoid bubbling and entry of leaves, twigs or debris into 107 the sampling bottle. Some of the parameters were analysed in the field immediately after 108 collection of samples. Then the water samples were directly taken to the lab and analysed for 109 various physico-chemical parameters. Samples were analysed for following physico-chemical 110 parameters viz. temperature, turbidity, total solids, total dissolved solids, total suspended 111 solids, pH, total hardness (EDTA Titration method), calcium hardness, magnesium hardness, 112 total alkalinity (by simple titration method), chloride acidity, dissolved oxygen (Winkler 113 114 method), biochemical oxygen demand (5 days incubation method) and chemical oxygen 115 demand (by dichromate titration method).

116

117 Study area

The present study was performed on Malin river which is situated in Najibabad district Bijnor Uttar Pradesh. Najibabad is located at 29.63N, 78.33E; it has an elevation of 295 meter (1014 feet). Malin river is the principal source of water for agriculture and other activities. This river is formed by joining of many mountain springs in Garhwal region. It is non-perennial river, get partially dried in summer and it is flooded with water in monsoon. Thus the quantity of water available from river varies throughout the year. It normally decreases in summer when the demand for water is on peak. Malin River covers about 140-150km with a catchment area of about 400 km² through 3 district named Pauri Garhwal, Kotdwara and Bijnor. Malin River merges in the Ganga River at the RavalliGhat in the Bijnor city. The main activities responsible for Malin river water pollution are runoff from agricultural fields, domestic waste form the city and villages situated on the bank of river and effluent from Kishan Sahkari Sugar mill. All the sampling sites were shown in table 1 and figure 1.

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131 Calculating of Water Quality Index (WQI)

132 Calculating of water quality index is to turn complex water quality data into information that is understandable and useable by the public. Therefore, water Quality Index (WQI) is a very 133 134 useful and efficient method which can provide a simple indicator of water quality and it is based on some very important parameters. In current study, Water Quality Index (WQI) was 135 136 calculated by using the Weighted Arithmetic Index method as described by (Cude, 2001; 137 Brown et al., 1970). In this model, different water quality components are multiplied by a weighting factor and are then aggregated using simple arithmetic mean. For assessing the 138 139 quality of water in this study, firstly, the quality rating scale (Qi) for each parameter was 140 calculated by using the following equation;

- 141
- 142 Where,

143 Qi = Quality rating of ith parameter for a total of n water quality parameters

$$Qi = \frac{(V_{observed} - V_{ideal})}{(V_{standard} - V_{ideal})} x100$$

146 Videal = Ideal value of that water quality parameter can be obtained from the standard147 Tables.

148 Videal for pH = 7 and for other parameters it is equal to zero, but for DO Videal = 14.6 mg/L

149 Vstandard = Recommended WHO standard of the water quality parameter.

Then, after calculating the quality rating scale (Qi), the Relative (unit) weight (Wi) was calculated by a value inversely proportional to the recommended standard (Si) for the corresponding parameter using the following expression;

153

$$Wi = \frac{K}{Xi}$$

154 Where,

• Wi = Relative (unit) weight for nth parameter

156 • Xi= Standard permissible value for nth parameter

- **•** K= Proportionality constant.
- That means, the Relative (unit) weight (WI) to various water Quality parameters are inverselyproportional to the recommended standards for the corresponding parameters.

Finally, the overall WQI was calculated by aggregating the quality rating with the unit weightlinearly by using the following equation:

$$WQI = \frac{\Sigma QiWi}{\Sigma Wi}$$

163

- 164 Where,
- Qi = Quality rating
- Wi = Relative weight

In general, WQI is defined for a specific and intended use of water. In this study the WQI
was considered for human consumption or uses and the maximum permissible WQI for the
drinking water was taken as 100 score.

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172 **Results and Discussion**

The results of various physico-chemical parameters of River Malin are tabulated in table 3 and 4 and figure 2 and 3 while their WQI values are given in table 5.

175

176 Turbidity of water is an important factor that controls the energy relationship at different tropic levels. It is essentially a function of reflection of light from the surface and is 177 178 influenced by the absorption characteristics of both water and of its dissolved and particulate 179 matter. During the study period the monthly values of turbidity was ranged from 22.7 NTU to 83.3NTU. The minimum monthly average value of turbidity were found 27.2 NTU \pm 3.53 in 180 the month of June and maximum monthly average value of turbidity were found 70.1 NTU 181 182 ±16.32 in the month of August (Table-2 and Graph-1). Turbidity values are generally found higher in Monsoon period due to heavy rainfall in mountain areas of Kotdwara region, the 183 184 origin Point of Malin River. The annual values of turbidity were ranged from 35.7 NTU to 48.9 NTU and annual average was observed 43.9±15.56. A more or less same trend was 185 186 observed by Khanna et al. (2010).

187

The solids represent the total salts and dirts remain after a particular amount of water sample 188 evaporated. Ecological imbalance in the aquatic ecosystem was caused by technical abrasive 189 190 action of total solids. During the study period the monthly values of TS was ranged from 808.0 mg/l to 1094.7 mg/l. The minimum monthly average value of TS were found 864.0 191 192 $mg/l \pm 58.07$ in the month of May and maximum monthly average value of TS were found 193 1074.1 mg/l ±22.31 in the month of August (Table-2 and Graph-1). TS values are generally 194 found higher in Monsoon period due to heavy rainfall in mountainous areas of Kotdwara 195 region, the origin Point of Malin River. In rainy season when rain fall occurs the river flows 196 with a high velocity and caused soil erosion in nearby areas which increase the total solids in 197 river water. The annual average values of TS were ranged from 939.3 mg/l to 991.4 mg/l and 198 annual average values were observed 963.1±78.64. A more or less same trend was observed 199 by Bhutiani and Khanna (2005).

200

201 Total dissolved solids (TDS) comprise inorganic salts (principally calcium, magnesium, 202 potassium, sodium, bicarbonates, chlorides, and sulphates) and some small amounts of 203 organic matter that are dissolved in water. It signifies the inorganic pollution load of water 204 system. During the study period the monthly values of TDS was ranged from 512.0 mg/l to 205 746.3 mg/l. The minimum monthly average value of total dissolved solid were found 561.7 206 $mg/1 \pm 52.32$ in the month of May and maximum monthly average value were observed 207 714.8 mg/l \pm 22.12 in the month of August (Table-2 and Graph-1). The annual average values 208 of TDS were ranged from 623.7 mg/l to 642.1 mg/l and annual average were observed 635.1 209 mg/l ±55.31. A more or less same trend was observed by Khanna et al. (2014) and Bhutiani 210 et al. (2017).

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TSS was previously called non-filterable residue (NFR), but was changed to TSS because of ambiguity in other scientific disciplines. During the study period the monthly values of TSS was ranged from 271.0 mg/l to 391.7 mg/l. The minimum monthly average value of total suspended solid were found 297.0 \pm 15.68 mg/l in the month of June and maximum monthly average value were observed 359.3 mg/l \pm 24.91 in the month of August (Table-2and Graph-1). The annual average values of TSS were ranged from 305.4 mg/l to 350.4 mg/l and annual average were observed 327.9 mg/l \pm 24.0. A more or less same trend was observed by Khanna *et al.* (2014).

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222 The amount of DO present in surface waters depends on water temperature, turbulence, 223 salinity, and altitude Natural waters in equilibrium with the atmosphere will contain DO concentrations ranging from about 5 to 14.5 mg O2 per liter. The DO concentration present in 224 225 water reflects atmospheric dissolution, as well as autotrophic and heterotrophic processes 226 that, respectively, produce and consume oxygen. DO is the factor that determines whether biological changes are brought by aerobic or anaerobic organisms. Thus, dissolved-oxygen 227 measurement is vital for maintaining aerobic treatment processes intended to purify domestic 228 229 and industrial wastewaters. A rapid fall in the DO indicates a high organic pollution in the river (Shah and Joshi, 2017). During the study period the monthly values of Dissolved 230 Oxygen was ranged from 5.9 mg/l to 9.4 mg/l. The minimum monthly average value of 231 232 Dissolved Oxygen were found 7.2 mg/l ± 1.04 in the month of September and maximum 233 monthly average value were observed 8.0 mg/l ± 1.09 in the month of June (Table-3 and 234 Graph-2). The annual average values of Dissolved Oxygen were ranged from 6.3 mg/l to 8.7 mg/l and annual average were observed 7.5 mg/l ±0.27. A more or less same trend was 235 236 observed by Kumar et al. (2012) and Arya and Gupta (2013).

237

Biological oxygen Demand is a measure of oxygen in the water that is required by the aerobic 238 organisms to decompose the organic matter. During the study period the monthly values of 239 240 biological oxygen demand (BOD) was ranged from 3.7 mg/l to 26.0 mg/l. The minimum monthly average value of biological oxygen demand (BOD) were found $11.2 \text{ mg/l} \pm 9.98$ in 241 242 the month of May and maximum monthly average value were observed $13.0 \text{ mg/l} \pm 9.97$ in the 243 month of April (Table-3 and Graph-2). The annual average values of biological oxygen 244 demand (BOD) were ranged from 4.2 mg/l to 22.5 mg/l and annual average were observed 245 12.1 mg/l \pm 0.54. COD is an oxygen demand to decompose the biodegradable as well as non-246 biodegradable organic waste. COD pointing to a deterioration of water quality likely caused 247 by discharge of municipal waste water. During the study period the monthly values of 248 chemical oxygen demand (COD) was ranged from 5.3 mg/l to 63.6 mg/l. The minimum 249 monthly average value of chemical oxygen demand (COD) were found $33.2 \text{ mg/l} \pm 19.75$ in 250 the month of May and maximum monthly average value were observed 36.0 mg/l±24.59 in 251 the month of November (Table-3and Graph-2). An increase in the COD values was found in 252 winter because of sugar mill effluent mixing in the river water. The annual average values of 253 chemical oxygen demand (COD) were ranged from 12.0 mg/l to 61.8 mg/l and annual 254 average value were observed 35.2 mg/l ± 1.01 . A more or less same trend was observed by 255 Kumar et al. (2012) and Arya and Gupta (2013).

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Alkalinity is the name given to the quantitative capacity of water to neutralize an acid. During the study period the monthly values of Alkalinity was ranged from 99.1 mg/l to 136.7 mg/l. The minimum monthly average value of Alkalinity were found 118.9 mg/l \pm 5.55 in the month of November and maximum monthly average value were observed 122.1 mg/l \pm 5.88 in the month of March (Table-3 and Graph-2). The annual average values of Alkalinity were ranged from 114.0 mg/l to 126.4 mg/l and annual average value were observed 120.6 mg/l \pm 1.01. A more or less same trend was observed by Ruhela *et al.* (2017) and Bhutiani *et al.*, 264 (2017). During the study period the monthly values of chlorides was ranged from 21.7 mg/l 265 to 38.7 mg/l. The minimum monthly average value of chlorides were found 26.5 mg/l \pm 4.64 266 in the month of November and maximum monthly average value were observed 28.9 mg/l 267 \pm 6.82 in the month of April (Table-3 and Graph-2). The annual average values of chlorides 268 were ranged from 23.2 mg/l to 35.8 mg/l and annual average value were observed 27.9 mg/l 269 \pm 0.85. A more or less same trend was observed by Khanna *et al.* (2012).

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271 Total hardness (TH) is a parameter of water quality used to describe the effect of dissolved 272 mineral (Ca and Mg), determining solubility of water for domestic, industrial and drinking 273 purpose attributed to presence of bicarbonates, sulphate, chloride and nitrates of Calcium and Magnesium. During the study period the monthly values of total hardness (TH) was ranged 274 275 from 320.8 mg/l to 351.7 mg/l. The minimum monthly average value of total hardness (TH) 276 were found $337.8 \text{ mg/l} \pm 11.81$ in the month of March and maximum monthly average value 277 were observed 344.1 mg/l \pm 5.92 in the month of January (Table-3and Graph-2). The annual average values of total hardness (TH) were ranged from 339.2 mg/l to 343.7 mg/l and annual 278 279 average value were observed 341.0 mg/l \pm 1.84. A more or less same trend was observed by Bhutiani et al. (2017). The occurrence of calcium hardness (CaH) in water is mainly due to 280 the presence of lime stone, gypsum dolomite and gypsi-ferrous material. During the study 281 282 period the monthly values of calcium hardness (CaH) was ranged from 106.2 mg/l to 129.3 283 mg/l. The minimum monthly average value of calcium hardness (CaH) were found 114.0 284 $mg/1 \pm 8.41$ in the month of May and maximum monthly average value were observed 119.3 mg/l ±7.32 in the month of March (Table-3 and Graph-2). The annual average values of 285 286 calcium hardness (CaH) were ranged from 111.0 mg/l to 125.8 mg/l and annual average value 287 were observed 116.2 mg/l \pm 1.37. Magnesium ranked fourth after sodium in sea water. During the study period the monthly values of calcium hardness (CaH) was ranged from 106.2 mg/l 288 to 129.3 mg/l. The minimum monthly average value of calcium hardness (CaH) were found 289 290 114.0 mg/l \pm 8.41 in the month of May and maximum monthly average value were observed 291 119.3 mg/l ±7.32 in the month of March (Table-3 and Graph-2). The annual average values 292 of calcium hardness (CaH) were ranged from 111.0 mg/l to 125.8 mg/l and annual average 293 value were observed 116.2 mg/l \pm 1.37. Approximately similar trend were observed by Arya 294 and Gupta (2013).

295 296

297 Conclusion

298 The point sources contributing to river Malin have very high organic pollution deteriorating water quality of the river Malin. The river Malin is subjected to varying degree of pollution 299 300 caused by numerous untreated and/or partially treated waste inputs of municipal and 301 industrial effluents as assessed by water quality index also. Water quality index is an efficient 302 tool to classify the water of the river for their various advantageous uses and give a rapid and 303 precise idea about the pollution load in the river that may be worthwhile for policy makers. 304 On the basis of the present investigation, it was found that the water Malin river is not fit for 305 direct human consumption.

306

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- **309 Conflict of Interest**
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- 312

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degraded river in Asansol industrial area(West Bengal). *Nature, Environment and pollution technology*, 1(2), 181-189.

Table 1: Showing the sampling sites.

SN	Sampling Site	Co-ordinates
1	Malin River near Shahpur village- (Fig-2)	29.62N, 78.33E
2	Malin River near Basantimata palace- (Fig-3)	29.61N, 78.33E
3	Malin River near Alipura village- (Fig-4)	29.61N, 78.31E
4	Malin River near Kalheri village- (Fig-5)	29.61N, 78.29E

Table 2: Water Quality Index (WQI) and its status according to Chaterjee and Raziuddin (2002).

Water quality Index Level	Water Quality Status
0-25	Excellent water quality
26-50	Good water quality
51-75	Poor water quality
76-100	Very poor water quality
>100	Unsuitable for drinking





Figure 1: Map showing all the four sampling sites.



7 Figure 2: Showing sampling site-1 Malin River near Shahpur village.

